

Enantioselective 1,2-Difunctionalization of Dienes Enabled by Chiral Palladium Complex-Catalyzed Cascade Arylation/Allylic Alkylation Reaction

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S Supporting Information

ABSTRACT: A Pd-catalyzed highly enantioselective three-component coupling of 1,3-dienes with aryl iodides and sodium dialkyl malonates has been successfully established by using a H₈-BINOL-based phosphoramidite ligand. This reaction proceeded via a Pd-catalyzed cascade arylation and asymmetric allylic alkylation reaction, providing an efficient strategy for the enantioselective 1,2-difunctionalization of 1,3-dienes.

Difunctionalization of alkenes provides a wide range of structurally diverse functionalized chemicals that are of great importance in organic synthesis, and it has hence been considered a powerful strategy in synthetic organic chemistry.¹ 1,3-Dienes are easily accessible chemicals,² basically able to participate in a wide spectrum of reactions acting on the carbon-carbon double bonds. Indeed, the past several decades have shown that 1,3-diene derivatives are versatile reagents to render the invention of a large number of fundamentally important and synthetically significant methodologies, as exemplified by stereoselective cycloaddition reactions and polymerizations, which have shown widespread applications in the medicinal chemistry and material science.³ It has been recognized that the 1,3-dienes can undergo nucleopalladation with Pd(II) and one nucleophile (Nu₁⁻) to generate a π-allyl-Pd intermediate, which can then be trapped by another nucleophile (Nu₂⁻) to afford either 1,2- or 1,4-products, releasing Pd(0) that is oxidized into catalytically active Pd(II) for the next catalytic cycle (eq 1, Figure 1).^{4–7} Pd(0) complexes have also been identified to afford various 1,2-difunctionalization reactions upon undergoing oxidative addition with a high-oxidation-state compound and subsequent Heck insertion of a 1,3-diene to form a π-allyl-Pd

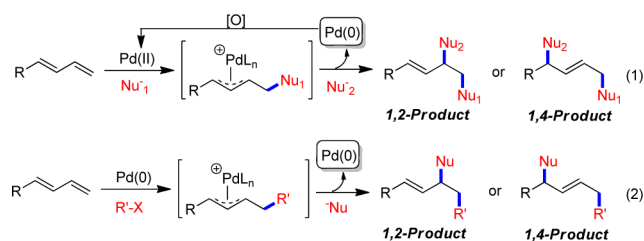


Figure 1. Pd-catalyzed difunctionalization of 1,3-dienes.

species, which ultimately reacts with a nucleophile to generate a 1,2- or 1,4-addition-like product (eq 2, Figure 1).^{8–12} However, among these transformations, highly enantioselective 1,2-difunctionalization reactions of 1,3-dienes, in particular, the protocols for the formation of two carbon-carbon bonds, have rarely been reported.^{10,11} Very recently, Sigman and co-workers established a Pd(0)-catalyzed intermolecular 1,2-diarylation reaction of 1,3-dienes with aryldiazonium salts and arylboronic acids, allowing the installation of two different aryl groups.¹² In the presence of a chiral bicyclo[2.2.2]octadiene ligand, the reaction was able to give a good enantiomeric excess (ee), but a rather low yield (Figure 2a). As a consequence, highly efficient and enantioselective 1,2-difunctionalization reactions of 1,3-

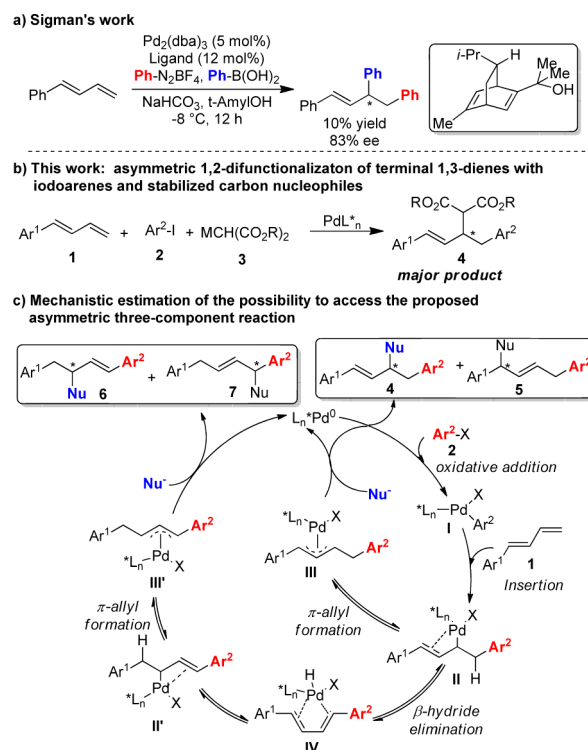


Figure 2. Pd-catalyzed asymmetric 1,2-difunctionalization of 1,3-dienes.

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dienes remain highly challenging and are in great demand. Herein, we will describe a highly regio- and stereoselective 1,2-difunctionalization of terminal 1,3-dienes with iodoarenes and stabilized carbon nucleophiles catalyzed by chiral Pd complexes (Figure 2b).

As shown in Figure 2c, the Pd(II) intermediate **I**, generated from oxidative addition of a Pd(0) complex to an aryl iodide **2**, undergoes a Heck insertion reaction to give an allylic Pd intermediate, **II**, which will be able to undergo isomerization to form a π -allyl-Pd intermediate, **III**^{8–12} and then to participate in allylic alkylation reaction with a stabilized carbon nucleophile, principally giving rise to either a 1,2-product, **4**, or a 1,4-product, **5**. Alternatively, the intermediate **II** would experience sequential events including β -hydride elimination, reinsertion reaction via Pd complex **IV**, and isomerization via Pd species **II'** to yield π -allyl-Pd intermediate **III'**, as indicated by Sigman.^{8p,q,9} The π -allyl-Pd species **III'** will also undergo the allylic alkylation reaction, to basically generate two regioisomers, **6** and **7**. Apparently, the simultaneous control of both the regio- and stereoselectivity renders the proposed three-component reaction much more challenging than similar reactions established already.^{8–11} Since the chiral ligands principally coordinates to all of the Pd species formed in whole reaction process, we believe that the use of chiral ligands will be able to provide solutions to formidable issues associated with selectivities. However, the requirement for the chiral ligands to allow the proposed three-component reaction proceeding smoothly is quite critical: they not only enable the Pd to smoothly undergo the oxidative addition to aryl iodide and the subsequent insertion reaction but also are able to efficiently control the stereoselectivity of the asymmetric allylic alkylation.

The feasibility of the hypothesis was initially explored by screening chiral ligands for the three-component reaction of (*E*)-1-phenylbutadiene (**1a**) with iodobenzene (**2a**) and sodium dimethyl malonate (**3a**) (Table 1). The chiral ligands commonly employed in the asymmetric allylic alkylation, including Trost ligands,¹³ chiral phosphino-oxazoline-type P,N ligands,¹⁴ and others, which were reported to deliver excellent levels of enantioselectivity in Pd catalysis,¹⁵ were initially evaluated. However, they were unable to give good results (Table S1, Supporting Information). Chiral phosphoramidite ligands that perform well in controlling the stereoselectivity of allylic substitutions¹⁶ were then examined. In the presence of 10 mol % of BINOL-based phosphoramidite **L1**,¹⁶ Pd(OAc)₂ was indeed able to catalyze the three-component reaction in THF at 80 °C, and the desired 1,2-aryllkylation product **4a** was isolated in 77% yield and with excellent regioselectivity, but without enantioselectivity (entry 1). Finely tuning the structure of BINOL-derived phosphoramidites revealed that ligands **L2**–**L7**, bearing substituents at the 3,3'-positions of the binaphthyl backbone and 7-position of 1,2,3,4-tetrahydroquinoline, enabled the reaction to deliver even more promising levels of asymmetric induction (up to 81% ee, entries 2–7). A little higher ee was provided by H₈-BINOL-based phosphoramidite **L8** (83% ee, entry 8). The examination of solvents found that the reaction performed well in ether solvents (entries 9–12), and the highest enantioselectivity of 87% ee was observed in MTBE (entry 9). The variation of reaction temperature was unable to significantly alter the stereoselectivity (entries 13 and 14), but a much diminished yield was given at lower temperature (entry 13). The Pd source did not show significant effect on the reaction (Table S2, Supporting Information).

Table 1. Optimization of the Reaction Conditions^a

entry	ligand	solvent	yield (%) ^b	ee (%) ^c	ratio 4a:5a ^d
1	L1	THF	77	0	>15:1
2	L2	THF	84	5	>15:1
3	L3	THF	21	11	>15:1
4	L4	THF	74	9	>15:1
5	L5	THF	82	74	>15:1
6	L6	THF	90	79	>15:1
7	L7	THF	73	81	>15:1
8	L8	THF	90	83	>15:1
9	L8	MTBE	89	87	>15:1
10	L8	DMF	trace	–	–
11	L8	DCM	5	74	>15:1
12	L8	MeCN	27	87	>15:1
13 ^e	L8	MTBE	38	88	>15:1
14 ^f	L8	MTBE	89	87	>15:1

^aUnless indicated otherwise, reactions of **1a** (0.10 mmol), **2a** (0.15 mmol), **3a** (0.50 mmol), Pd(OAc)₂ (0.005 mmol), and **L** (0.010 mmol) were carried out in a solvent (2 mL) at 80 °C for 72 h.

^bIsolated yields. ^cDetermined by HPLC analysis. ^dDetermined by ¹H NMR analysis. ^eThe reaction was carried out at 60 °C. ^fThe reaction was carried out at 100 °C.

After the optimal reaction conditions were established, the generality of the asymmetric transformation was subsequently examined. A variety of arylbutadienes (**1b**–**1h**) and aryl iodides (**2b**–**2h**), possessing the same aryl groups, were first examined in the presence of 5 mol% of Pd(OAc)₂ and 10 mol% of **L8** in MTBE at 80 °C (Figure 3). It is necessary to mention that the structures of **4** and **5** are respectively identical to those of **6** and **7** (Figure 2c) in these cases, and thus only two different regioisomers are generated. The presence of either electron-releasing or deficient substituent at the *para*-position was nicely tolerated and highly regioselectively generated the target products in high yields and enantioselectivity (**4b**–**4d**). Obviously, the substitution pattern had considerable effect on both regio- and stereoselectivities (**4b**, **4e**, and **4f**). For instance, an excellent ee but a moderate regioselection was obtained in the reaction involving 2-methylphenylbutadiene and 2-methyliodobenzene (**4f**). In contrast, a little lower ee value, but an excellent regioselection was given to the case with either *p*- or *m*-methylphenyl substrate (**4b** or **4e**). Comparing with the results of the substrate with a methyl group at the *ortho*-position (**4f**), the presence of an electron-withdrawing group such as chloride at the *ortho*-position of aryl diene and aryl iodide led to high regio- and enantioselectivities and a good yield (**4g**). Further, the sterically hindered naphthyl substrates also underwent a smooth three-component coupling reaction in a satisfactory yield and with high levels of regio- and enantioselectivities (**4h**). Moreover, different malonates were also examined, providing excellent yields (**4i**, 90% yield, and **4j**, 88% yield) and high levels of enantioselectivities (**4i**, 83% ee, and **4j**, 85% ee).

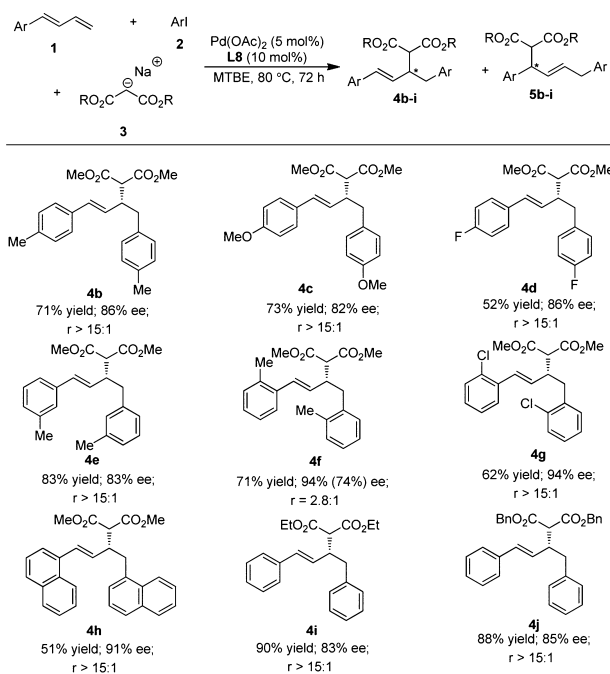


Figure 3. Scope of arylbutadienes and aryl iodides containing the same aryl groups. Reactions of **1a** (0.10 mmol), **2a** (0.15 mmol), **3a** (0.50 mmol), Pd(OAc)₂ (0.005 mmol), and L8 (0.010 mmol) were carried out in MTBE (2 mL) at 80 °C for 72 h. Yields are reported as a mixture of regioisomers of **4** and **5**. The ee values were determined by HPLC analysis. The ratio (*r*) of **4:5** was determined by ¹H NMR analysis. The number in the parentheses for **4f** is the ee value of the minor regiomers **5f**; for characterization of **5f**, see Supporting Information.

Next, the generality for a number of arylbutadienes **1** and aryl iodides **2**, which contain different aryl substituents, was investigated (Figure 4). Basically, four different regioisomers are generated from the reaction, and hence will bring even more challenges to the simultaneous control of both the regio- and stereoselectivities. To our delight, the application of the optimal conditions to the three-component reaction of (*E*)-1-phenylbutadiene (**1a**) with different iodobenzenes **2** was quite successful to give the corresponding chiral products **4k–4r** in high yields ranging from 74% to 93% and with excellent levels of enantioselectivity of up to 98% ee. Notably, the regioselectivity to preferentially generate **4** over other regioisomers was nicely controlled (**4k–4q**) while less regioselective control of 1,2-addition was observed for 2-methyliodobenzene reacting with (*E*)-1-phenylbutadiene to afford **4r**. However, the regiometric ratios of 1,2-addition product to other regioisomers seemingly suffer from the *ortho*- or *meta*-substitution with an electron-releasing group (**4p** and **4r** vs **4q**). Using iodobenzene (**2a**) as the cross coupling partner, a variety of arylbutadienes were finally evaluated. Obviously, both the regio- and enantioselectivities are highly sensitive to the aryl substituents (**4s–4u**). For example, low regioselectivity was obtained for the reaction with *ortho*-substituted phenylbutadiene, albeit with a high enantioselectivity (**4s**), while 2-thiophenyl- and 1-naphthyl-substituted butadienes could lead to 1,2-products with high regio- and enantioselection (**4t** and **4u**). The structure and absolute configuration of product **4m** was assigned by X-ray analysis (see Supporting Information).

As shown in Figure 2c, all of the regioisomers are generated from the nucleophilic substitution of two π -allyl-Pd intermediates **III** and **III'**. To understand how the π -allyl-Pd species are generated,

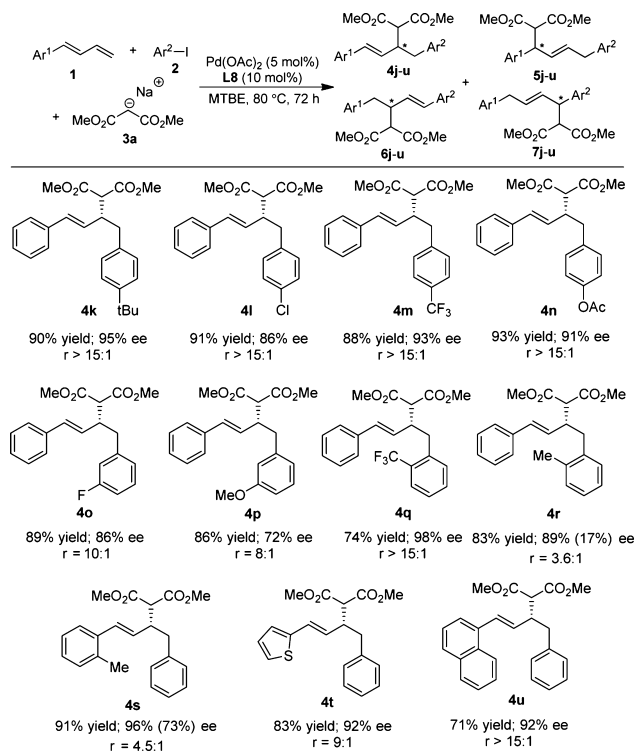
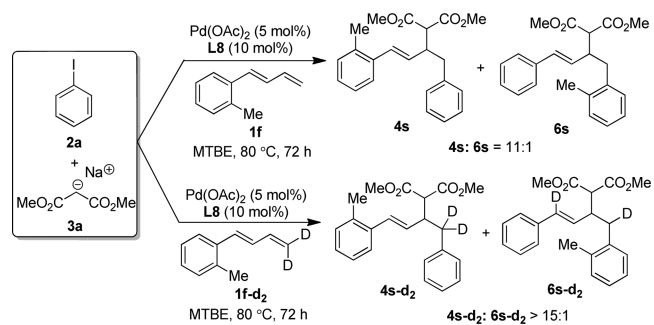


Figure 4. Scope of arylbutadienes and aryl iodides containing different aryl groups. Reactions of **1** (0.10 mmol), **2** (0.15 mmol), **3a** (0.50 mmol), Pd(OAc)₂ (0.005 mmol), and L8 (0.010 mmol) were carried out in MTBE (2 mL) at 80 °C for 72 h. Yields are reported as a mixture of regioisomers of **4**, **5**, **6**, and **7**. The ee values of major products **4** were determined by HPLC analysis. Unless stated otherwise, the ratio (*r*) represents major products (**4**):regiomers (**5+6+7**), determined by ¹H NMR analysis. The numbers in the parentheses for **4r** and **4s** are the ee values of regiomers **6r** and **6s**, respectively, determined by HPLC analysis. The ratio of **4r:6r:(5r+7r)** is 1.0:0.10:0.18. The ratio of **4s:6s:(5s+7s)** is 1.0:0.09:0.13. For the characterization of the minor regiomers **6s**, see Supporting Information.

experimental studies on the isotope effect were conducted by the reaction of *d*₂-*o*-methylphenylbutadiene (**1f-d**₂), iodobenzene (**2a**), and sodium dimethyl malonate (**3a**). A similar isotope labeling experiment, reported by Sigman, identified that no deuterium migration occurs.^{8p} However, the deuterium migration was found in the minor regiomers **6s-d**₂ (Scheme 1). More interestingly, an obvious isotope effect was observed for the regioselectivity. An enhanced regioselectivity (>15/1 vs 11/1) was obtained for the *d*₂-*o*-methylphenylbutadiene (**1f-d**₂), probably due to the fact that the β -deuteride elimination and reinsertion reactions proceed much more slowly than the similar

Scheme 1. Experimental Studies on Isotope Effect



transformations with **1f**,¹⁷ thereby causing the allyl-Pd intermediate **III'** to be formed much more slowly than **III**. These experimental results, in aggregate, indicated that the β -hydride elimination and reinsertion reaction to form allyl-Pd intermediate **II'** indeed occurred, leading to minor products **6** and **7** (Figure 2c).

In summary, we have established a chiral Pd complex-catalyzed asymmetric three-component reaction for the difunctionalization of 1,3-dienes with aryl iodides and sodium dialkyl malonate, resulting in high yields and excellent levels of regio- and enantioselectivities. The H₈-BINOL-based phosphoramidite turned out to be the optimal ligand, which not only provides high catalytic activity but also is able to efficiently control the regio- and stereoselectivity. The reaction proceeds via a Pd-catalyzed cascade arylation and asymmetric allylic alkylation, capable of tolerating a broad scope of substrates, including 1,3-dienes and iodoaryl compounds. Two different types of π -allyl-Pd intermediates, respectively generated from migratory insertion of the 1,3-diene to aryl-Pd(II) and sequential events, including Heck insertion, β -hydride elimination, and reinsertion reactions, are both involved in the whole reaction process. Notably, this protocol actually provides an important alternative strategy for the enantioselective difunctionalization of 1,3-dienes, leading to synthetically useful chiral chemicals that were hard to prepare via the classical asymmetric allylic alkylation.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b08734.

Experimental procedures; compound characterization data (PDF)

X-ray data of **4m** (CIF)

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Author Contributions

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) (a) Sigman, M. S.; Werner, E. W. *Acc. Chem. Res.* **2012**, *45*, 874. (b) Jensen, K. H.; Sigman, M. S. *Org. Biomol. Chem.* **2008**, *6*, 4083. (c) McDonald, R. I.; Liu, G.; Stahl, S. S. *Chem. Rev.* **2011**, *111*, 2981. (d) Schultz, D. M.; Wolfe, J. P. *Synthesis* **2012**, *44*, 351.
- (2) For a recent review on diene synthesis, see: De Paolis, M.; Chataigner, I.; Maddaluno, J. *Top. Curr. Chem.* **2012**, *327*, 87.
- (3) For selected reviews, see: (a) Mackay, E. G.; Sherburn, M. S. *Synthesis* **2015**, *47*, 1. (b) Eschenbrenner-Lux, V.; Kumar, K.; Waldmann, H. *Angew. Chem., Int. Ed.* **2014**, *53*, 11146. (c) Porri, L.; Giarrusso, A. *Comprehensive Polymer Science*; Pergamon Press: Oxford, 1989; Vol. 4, p 53. (d) Taube, R.; Sylvester, G. *Applied Homogeneous Catalysis with Organometallic Compounds*; Wiley-VCH: Weinheim, 1996; Vol. 1, p 280.
- (4) Bäckvall, J. E. In *Metal-Catalyzed Cross-Coupling Reactions and More*; De Meijere, A., Bräse, S., Oestreich, M., Eds.; Wiley-VCH: Weinheim, 2014; p 875.
- (5) (a) Bar, G. L. J.; Lloyd-Jones, G. C.; Booker-Milburn, K. I. *J. Am. Chem. Soc.* **2005**, *127*, 7308. (b) Houlden, C. E.; Bailey, C. D.; Ford, J. G.; Gagné, M. R.; Lloyd-Jones, G. C.; Booker-Milburn, K. I. *J. Am. Chem. Soc.* **2008**, *130*, 10066. (c) Coscia, R. W.; Lambert, T. H. *J. Am. Chem. Soc.* **2009**, *131*, 2496. (d) Widenhoefer, R. A. *Angew. Chem., Int. Ed.* **2009**, *48*, 6950. (e) Xing, D.; Yang, D. *Org. Lett.* **2013**, *15*, 4370.
- (6) Liao, L.; Sigman, M. S. *J. Am. Chem. Soc.* **2010**, *132*, 10209.
- (7) For selected examples, see: (a) Bäckvall, J. E.; Bystroem, S. E.; Nordberg, R. E. *J. Org. Chem.* **1984**, *49*, 4619. (b) Bäckvall, J. E.; Nystroem, J. E.; Nordberg, R. E. *J. Am. Chem. Soc.* **1985**, *107*, 3676. (c) Bäckvall, J. E.; Andersson, P. G. *J. Am. Chem. Soc.* **1990**, *112*, 3683. (d) Bäckvall, J. E.; Andersson, P. G. *J. Am. Chem. Soc.* **1992**, *114*, 6374. (e) Andersson, P. G.; Bäckvall, J. E. *J. Am. Chem. Soc.* **1992**, *114*, 8696. (f) Castano, A. M.; Bäckvall, J.-E. *J. Am. Chem. Soc.* **1995**, *117*, 560. (g) Banerjee, D.; Junge, K.; Beller, M. *Org. Chem. Front.* **2014**, *1*, 368.
- (8) For selected examples, see: (a) Patel, B. A.; Dickerson, J. E.; Heck, R. F. *J. Org. Chem.* **1978**, *43*, 5018. (b) Patel, B. A.; Kao, L. C.; Cortese, N. A.; Minkiewicz, J. V.; Heck, R. F. *J. Org. Chem.* **1979**, *44*, 918. (c) Stakem, F. G.; Heck, R. F. *J. Org. Chem.* **1980**, *45*, 3584. (d) Fischetti, W.; Mak, K. T.; Rheingold, A. L.; Heck, R. F.; Stakem, F. G.; Kim, J. *J. Org. Chem.* **1983**, *48*, 948. (e) O'Connor, J. M.; Stallman, B. J.; Clark, W. G.; Shu, A. Y. L.; Spada, R. E.; Stevenson, T. M.; Dieck, H. A. *J. Org. Chem.* **1983**, *48*, 807. (f) Uno, M.; Takahashi, T.; Takahashi, S. *J. Chem. Soc., Chem. Commun.* **1987**, 785. (g) Grigg, R.; Sridharan, V.; Sukirthalingam, S.; Worakun, T. *Tetrahedron Lett.* **1989**, *30*, 1139. (h) Larock, R. C.; Fried, C. A. *J. Am. Chem. Soc.* **1990**, *112*, 5882. (i) Larock, R. C.; Berrios-Pena, N. G.; Narayanan, K. J. *J. Org. Chem.* **1990**, *55*, 3447. (j) Larock, R. C.; Berrios-Pena, N. G.; Fried, C. A.; Yum, E. K.; Tu, C.; Leong, W. *J. Org. Chem.* **1993**, *58*, 4509. (k) Kagechika, K.; Ohshima, T.; Shibasaki, M. *Tetrahedron* **1993**, *49*, 1773. (l) Ohshima, T.; Kagechika, K.; Adachi, A.; Sodeoka, M.; Shibasaki, M. *J. Am. Chem. Soc.* **1996**, *118*, 7108. (m) Deagostino, A.; Prandi, C.; Venturello, P. *Org. Lett.* **2003**, *5*, 3815. (n) Yoshida, M.; Sugimoto, K.; Ihara, M. *Org. Lett.* **2004**, *6*, 1979. (o) Yeh, M.-C. P.; Tsao, W.-C.; Tu, L.-H. *Organometallics* **2005**, *24*, 5909. (p) Liao, L.; Jana, R.; Urkalan, K. B.; Sigman, M. S. *J. Am. Chem. Soc.* **2011**, *133*, 5784. (q) McCammant, M. S.; Liao, L.; Sigman, M. S. *J. Am. Chem. Soc.* **2013**, *135*, 4167.
- (9) Saini, V.; O'Dair, M.; Sigman, M. S. *J. Am. Chem. Soc.* **2015**, *137*, 608.
- (10) Zhu, Y.; Cornwall, R. G.; Du, H.; Zhao, B.; Shi, Y. *Acc. Chem. Res.* **2014**, *47*, 3665.
- (11) (a) Han, J. W.; Hayashi, T. *Tetrahedron: Asymmetry* **2010**, *21*, 2193. (b) Flubacher, D.; Helmchen, G. *Tetrahedron Lett.* **1999**, *40*, 3867.
- (12) Stokes, B. J.; Liao, L.; de Andrade, A. M.; Wang, Q.; Sigman, M. S. *Org. Lett.* **2014**, *16*, 4666.
- (13) (a) Trost, B. M.; Van Vranken, D. L. *Angew. Chem., Int. Ed. Engl.* **1992**, *31*, 228. (b) Trost, B. M.; Van Vranken, D. L.; Bingel, C. *J. Am. Chem. Soc.* **1992**, *114*, 9327. (c) Trost, B. M. *Acc. Chem. Res.* **1996**, *29*, 355.
- (14) (a) Sprinz, J.; Helmchen, G. *Tetrahedron Lett.* **1993**, *34*, 1769. (b) Allen, J. V.; Coote, S. J.; Dawson, G. J.; Frost, C. G.; Martins, C. J.; Williams, J. M. J. *J. Chem. Soc., Perkin Trans. 1* **1994**, *15*, 2065. (c) Helmchen, G.; Pfaltz, A. *Acc. Chem. Res.* **2000**, *33*, 336. (d) Williams, J. M. J. *Synlett* **1996**, 705.
- (15) For reviews, see: (a) Trost, B. M.; Van Vranken, D. L. *Chem. Rev.* **1996**, *96*, 395. (b) Trost, B. M.; Crawley, M. L. *Chem. Rev.* **2003**, *103*, 2921. (c) Lu, Z.; Ma, S.-M. *Angew. Chem., Int. Ed.* **2008**, *47*, 258.
- (16) (a) Teichert, J. F.; Feringa, B. L. *Angew. Chem., Int. Ed.* **2010**, *49*, 2486. (b) Zhuo, C.-X.; Zheng, C.; You, S.-L. *Acc. Chem. Res.* **2014**, *47*, 2558.
- (17) (a) Ozawa, F.; Yamamoto, A.; Ito, T. *J. Am. Chem. Soc.* **1980**, *102*, 6457. (b) Gómez-Gallego, M.; Sierra, M. A. *Chem. Rev.* **2011**, *111*, 4857.